

Improving the CO₂ capture and conversion potential of MOF UiO-66 through sustainable creation of defects

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The daily release of high levels of carbon dioxide (CO₂) into the atmosphere is a serious threat for the sustainable future of the Earth, contributing to catastrophic climate changes. The capture of CO₂ and, subsequent, its conversion into added-value chemicals, is a valuable and cost-effective solution for the reduction of the CO₂ levels. However, bifunctional materials that can conciliate CO₂ capture and conversion simultaneously under sustainable conditions are scarce and needed.^{1, 2} Metal-organic frameworks (MOFs) are highly crystalline and porous structures formed by metallic center interconnected with organic ligands. A great number of MOFs have shown promising results in either CO₂ capture or conversion, for example, the cycloaddition of CO₂ with epoxides into cyclic carbonates.^{2, 3} Defect engineering, i.e. the deliberate formation of defects in MOFs structures increases the number of active sites, enhancing catalytic efficiency. UiO-66 is a strong candidate for defect induction, since its framework has high structural integrity even containing uncoordinated centers.^{3, 4}

In this work, we successfully synthesized a series of defective UiO-66-based materials to enhance the catalytic activity of the parent UiO-66, which were then tested for their ability to capture CO₂. The defects and improved catalytic potential were achieved through a mixed-ligand strategy, based on the mixture of different BDC-based ligands to introduce amine and halogen groups, that theoretically facilitate the opening of the epoxide rings. The materials were synthesized via conventional (solvothermal) and microwave-assisted methods, with the latter demonstrating a significantly lower environmental impact, as confirmed by an LCA study. Characterization techniques such as FTIR, PXRD, SEM-EDS, N₂ sorption, NMR and TGA were employed to confirm the engineered defects. Preliminary catalytic studies to evaluate the MOF performance in the cycloaddition of epichlorohydrin with CO₂ were also performed. This approach envisioned the enhancement of the catalytic performance of UiO-66 in the cycloaddition of CO₂ with epoxides to produce cyclic carbonates, without significant loss of CO₂ adsorption-desorption abilities and through a “greener” synthetic process.

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